Synthesis and magnetic properties of FeNi₃/Al₂O₃ core-shell **nanocomposites**

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Abstract. In this study, $FeNi₃/Al₂O₃$ core-shell nanocomposites, where individual $FeNi₃$ nanoparticles were coated with a thin layer of alumina, were fabricated by a modified sol-gel method. Several physical characterizations were performed on the samples of FeNi3/Al2O³ nanocomposites with different thickness of A_2O_3 shell. The encapsulation of FeNi₃ nanoparticles with alumina stops FeNi₃ agglomeration during heat treatment, and prevents interaction among the closely spaced magnetic FeNi₃ nanoparticles. The Al_2O_3 insulating shell improves the soft magnetic properties of FeNi3. The study of the complex permeability of the samples shows that the real part μ' of the permeability of the sample with Al molar content of 20% (Al/(Fe+Ni)) is as high as 12, and independent of frequency up to at least 1 GHz. The tunneling magnetoresistance arising from the presence of the Al_2O_3 shell have also been studied.

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1 Introduction

Magnetic nanocomposites, consisting of magnetic metal nanoparticles coated with a nonmagnetic insulator (e.g., polymer, alumina, or silica), have attracted particular attention due to their potential uses as microwave absorbing and shielding materials [1] as well as their applications in electromagnetic devices [2]. For example, nanocomposites with soft magnetic properties can be incorporated in high-frequency devices and in micro-transductors which require large magnetic moment and susceptibility, low coercivity, and high resistivity [3–5]. FeNi alloy is a well-known and commonly used soft magnetic material. However, due to the intrinsic property of the metal particles, the generation of eddy current severely limits its application at high frequency. A strategy to solve the problem is to cover the soft magnetic metallic nanoparticles by an inorganic and non-magnetic coating to create the core-shell nanostructure. This core-shell structure maintains the magnetic properties of metal and its alloys. And better performance can be achieved at high frequency when an insulating shell is applied. Recently, the magnetic and high-frequency properties of FeNi and other Fe-alloy nanocomposites with different insulator matrixes [6–9] have been reported. However, only a few are devoted to Al_2O_3 -based nanocomposites, and most are related to magnetic metals or alloys embedded in insulator matrixes which were not core-shell structures. Alumina is an effective material to form good barrier for spinpolarized tunnel [10]. The magnetic nanoparticles with

alumina shell would result in favorable magnetic properties that could be manipulated by controlling the thickness of the alumina shell. Also, by coating FeNi particles with alumina, one can increase the resistivity and hence reduce losses due to eddy currents. In this paper, we have successfully engineered a modified sol-gel approach to fabricate Al_2O_3 -coated FeN_i nanoparticles. The method we used is effective and the process is easy to control. We have also investigated the influences of alumina thickness on the magnetic and transport properties of FeNi₃ nanoparticles.

2 Experimental

Analytical grade reagents ferrous chloride $(FeCl₂·4H₂O)$, anhydrous aluminum chloride (AlCl3), nickel chloride (NiCl2*·*6H2O), citric acid monohydrate and ethanol absolute were used. The thickness of Al_2O_3 shell was controlled by the molar percentage of aluminum with respect to the total content of Fe and Ni. The total aluminum contents in the final as-prepared composites were 0% , 4% , 8% and 20% (Al/(Fe+Ni) molar ratio), and the corresponding samples are denoted as $FENi₃/Al₂O₃ - 0$, $FENi₃/Al₂O₃ - 4$, $FeNi₃/Al₂O₃ - 8$, and $FeNi₃/Al₂O₃ - 20$ hereafter. In a typical experiment, 0.01 mol FeCl2*·*4H2O, 0.03 mol NiCl₂·6H₂O and 0.09 mol citric acid monohydrate were dissolved in 100 ml ethanol. After stirring at 60 ◦C for 6 h, a clear sol was obtained. Then $0.0016 \;\mathrm{mol}$ AlCl₃ was added to the sol and the mixture was stirred for 4 h. After drying the sol at 80 ◦C, the xerogel was calcinated in air

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Fig. 1. X-ray diffraction patterns of $FeNi₃/Al₂O₃$ nanocomposites with different contents of Al.

at 400 \degree C for 3 h, followed by reduction at 800 \degree C in a hydrogen stream of 2 L/h for 4 h. Then the sample with 4 mol% Al/(Fe+Ni) (i.e. $FeNi₃/Al₂O₃$ -4) was obtained.

The phase identification and structure analysis of the as-prepared powder samples were characterized by X-ray powder diffraction (XRD) with Cu -K α radiation (Model D/Max-RA, Rigaku, Japan). The morphology of the samples was observed via transmission election microscopy (TEM) (Model JEM-200 CX, JEOL, Japan). The magnetic property was measured by a vibrating sample magnetometer (VSM) (Lakeshore, USA). For complex permeability and magnetoresistance measurements, the powder samples were pressed into a ring and a flake, respectively. Complex permeability spectra were measured with an impedance analyzer (Agilent4284A from 20 Hz to 1 MHz) and an impedance/material analyzer (Agilent4191B from 1 MHz to 1.8 GHz). The magnetoresistance measurement was conducted by the standard fourpoint method.

3 Results and discussion

The XRD patterns of three samples with different contents of Al are shown in Figure 1. The diffraction patterns of the three samples match well with the cubic structure of FeNi₃. There are no detectable peaks of metallic iron and nickel or their oxides. The results indicate the successful formation of the FeNi₃ alloy. It should also be noted that there is no evidence for the presence of crystalline Al_2O_3 either in the XRD patterns or in the electron diffraction patterns for all the samples. As a result, we deduce that the alumina in the $FeNi₃/Al₂O₃$ composites is in an amorphous state.

A typical TEM micrograph of the $FeNi₃/Al₂O₃$ nanocomposites with Al molar ratio of 8% is shown in Figure 2. The interface between the FeNi₃ cores and the $Al₂O₃$ shell can be observed clearly with high contrast. As indicated, the alumina shell is about 4 nm in thickness. The TEM image demonstrates that the core-shell

Fig. 2. TEM image of the nanoparticles of $FeNi₃/Al₂O₃ - 8$.

Fig. 3. Average grain size of FeNi₃ core and thickness of alumina shell as a function of Al molar concentration.

nanocomposites of $FeNi₃/Al₂O₃$ have been successfully fabricated by the simple chemical method.

The molar ratio between Fe and Al ions was found to be an important parameter for the determination of the size of the FeNi₃ core. As shown in Figure 3, the thickness of the alumina shell increases with increasing aluminum content. After reduction at the same temperature, the FeNi³ particles with thicker alumina shell are obviously smaller in size than those with thinner alumina shell. This indicates that the Al_2O_3 coating could hinder the agglomeration of the FeNi₃ nanoparticles to a certain extent.

The magnetic hysteresis loops at room temperature of the nanoparticles are shown in Figure 4. They are typical hysteresis loops of soft magnetic materials. It should be stressed that the magnetization is expressed in units of emu per gram of powder, with the mass of the nonmagnetic shell being included. Using the molar percentage of the FeNi₃ cores in the nanocomposites, the magnetizations of FeNi³ cores were derived. It was found that they were close to the value of FeNi₃ alloy without alumina layer. The alumina shell also has an effect on the coercivity of the FeNi₃ core. As shown in the inset of Figure 4, the coercivity increases obviously after the FeNi₃ particles were coated with alumina, and the coercivity increases slightly with increase in alumina content. As pointed out before, at equal synthesis temperature, the size of FeNi₃ particles with thicker A_2O_3 coating was smaller than that with the thinner coating. It's known that the coercivity increases with decreasing grain size for multidomain particles [11],

Fig. 4. Hysteresis loops for samples with different thickness of $Al₂O₃$ shell. The inset on the top left corner represents the influence of the Al_2O_3 shell thickness on the coercivity of Al_2O_3 coated FeNi₃ nanoparticles.

so it is reasonable to observe that smaller FeNi₃ nanoparticles exhibited higher coercivity. The change in H*^C* values in the alumina-coated samples confirms that the alumina coating is essential to avoid sintering of the FeNi₃ particles.

The frequency dependence of the complex permeability for all the samples is shown in Figure 5. It is observed that the real part of permeability. μ' of the uncoated sample decreased rapidly at 1 MHz. When the alloy was coated by alumina, the cut-off frequency increases obviously. For the sample of $\text{FeNi}_3/\text{Al}_2\text{O}_3\text{-}20$, μ' is independent of frequency up to at least 1 GHz. It is clear that the insulating Al_2O_3 shell improves the performance of the soft magnetic FeNi₃ cores at high frequency. The performance of conventional soft magnetic materials is strongly dependent on the operating frequency. For metallic magnetic materials, the low cut-off frequency is usually attributed to the eddy currents arising from poor insulation between particles. These currents can lead to significant losses when the applied field is alternating current (AC), particularly at high frequency, and can cause significant heating of the material. If there is no insulated shell, direct metal contact would lead to eddy current, which would cause μ' to decrease fast with frequency and the imaginary part μ'' reach a maximum at a lower frequency. But in the $\rm Al_2O_3\text{-}coated$ FeNi₃ structure, such limitations could be eliminated. As shown in Figure 5, with the increase in alumina content, the shell thickens and is compact enough for efficient insulation. The thickening of the Al_2O_3 coating enhances the resistivity of materials, consequently reducing the effect of eddy currents at high frequency as well as keeping the μ' almost constant and the imaginary part very small.

In magnetic nanoparticles, coupling between particles can be induced through direct exchange coupling and dipolar interaction. The former is the main way to improve the soft magnetic properties in permeability [6]. In the $FeNi₃/Al₂O₃$ nanocomposites, the dipolar interaction is the main effect because the magnetic particles are well

separated by the alumina coating, and exchange interactions are negligible. That is the reason why the μ' is not very high. On the other hand, compared with the sintered one, the pressed ring is lower in density. The possible reason could be that the nonmagnetic impurities such as boundaries and cavities from the low density may play an important role and may cause further decline in permeability. In other words, the permeability can be significantly improved by manipulating the coating thickness and the density.

 Al_2O_3 is an effective material to form good barrier for spin-polarized tunneling [10]. Since there is insulating alumina shell between the FeNi₃ particles, they may exhibit similar magneto-transport property as that in Ferromagnetic/Insulator/Ferromagnetic (FM/I/FM) tunnel junction, which exhibits tunneling magnetoresistance (TMR) effect. We conducted magnetoresistance measurements for the $FeNi₃/Al₂O₃ - 8$ and $FeNi₃/Al₂O₃ - 20$ samples at room temperature and 4 K, respectively. Resistance at zero field (R_0) was used as the reference when calculating MR percentage, and detailed investigation was performed between *±*12kOe (corresponding resistance is denoted by R_H). Figure 6 shows the resistance ratio [MR ratio = $(R_H - R_0)/R_0$] versus magnetic field curve. For the FeNi₃/Al₂O₃-8 and FeNi₃/Al₂O₃-20 samples, the magnetoresistance ratios were 0.6%, 1.1% at room temperature and 1.0%, 2.1% at 4 K, respectively. It has been pointed out that tunneling magnetoresistance is related intimately with the microstructure of the nanocomposite, such as the grain size, shape, size distribution, and the insulating shell structure [12,13]. If the coating is too thin, the cores of the particles can be in direct contact, leading to short circuit and decrease in tunneling effect; as a result, TMR is lower. This observation is in consistent with the results of the complex permeability measurements. At lower field over our samples, we found the little positive magnetoresistance effect which may be caused by anisotropic magnetoresistance (AMR) of FeNi₃ cores. Since Fe-Ni alloy has positive AMR at room temperature and its saturation field is low, TMR of the $FeNi₃/Al₂O₃$ nanocomposites cannot compare with AMR at the low field. However at higher field, the TMR of FeNi₃ increased and exceeded the AMR.

4 Conclusions

In summary, we have devised a simple chemical method for the synthesis of magnetic FeNi₃ nanoparticles that are enclosed by an alumina shell. The surface alumina layer prevents interaction between the closely spaced magnetic FeNi₃ particles and hinders grain growth and agglomeration during heat treatment. There is a decrease in M*^S* and increase in H_C with the increase of the $Al_2O_3/FeNi_3$ ratio. Moreover, Al_2O_3 insulating shell improves the soft magnetic properties of FeNi³ alloy. We could obtain soft magnetic nanoparticles with different cut-off frequency by controlling the thickness of the alumina shell. The study of complex permeability of the samples shows that the real part μ' of the permeability of the sample with Al molar

Fig. 5. The permeability spectra of the samples: (a) $F \text{e} \text{Ni}_3/\text{Al}_2\text{O}_3$ -0, (b) $F \text{e} \text{Ni}_3/\text{Al}_2\text{O}_3$ -4, (c) $F \text{e} \text{Ni}_3/\text{Al}_2\text{O}_3$ -8, and (d) $FeNi₃/Al₂O₃ - 20.$

Fig. 6. Magnetoresistances versus applied field for the samples (a) FeNi₃/Al₂O₃-8, (b) FeNi₃/Al₂O₃-20 at room temperature and 4 K.

content of 20% (Al/(Fe+Ni)) is independent of frequency up to at least 1 GHz, and the imaginary part of the permeability is very small. The MR effects of $Al_2O_3/FeNi_3$ nanocomposites at low field could be related to TMR arising from the presence of insulating alumina shell and the AMR of Fe-Ni alloy core.

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